# First-principles LDA+U and GGA+U study of plutonium oxides

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The electronic structure and properties of  $PuO_2$  and  $Pu_2O_3$  have been studied from first principles by the all-electron projector-augmented-wave (PAW) method. The local density approximation (LDA)+U and the generalized gradient approximation (GGA)+U formalism have been used to account for the strong on-site Coulomb repulsion among the localized Pu 5f electrons. We discuss how the properties of  $PuO_2$  and  $Pu_2O_3$  are affected by the choice of U as well as the choice of exchange-correlation potential. Also, oxidation reaction of  $Pu_2O_3$ , leading to formation of  $PuO_2$ , and its dependence on U and exchange-correlation potential have been studied. Our results show that by choosing an appropriate U it is promising to correctly and consistently describe structural, electronic, and thermodynamic properties of  $PuO_2$  and  $Pu_2O_3$ , which enables it possible the modeling of redox process involving Pu-based materials.

### I. INTRODUCTION

Plutonium dioxide  $(PuO_2)$  and sesquioxide  $(Pu_2O_3)$  are the only observed stoichiometric compounds formed at the surface of the metallic plutonium when exposed to dry air<sup>1</sup> (nonstoichiometric oxide may form by reaction of dioxide with water<sup>2</sup>). From this sense, plutonium corrosion and oxidation are often treated as equivalent topic. The plutonium corrosion plays a key role in considering the nuclear stockpile and storage of surplus plutonium. Therefore, a thorough understanding of the physical and chemical properties of the plutonium oxide is always needed.

From basic point of view, it can be visualized that many physical and chemical properties of the plutonium oxide are closely related to the quantum process of localization and delocalization for Pu 5f electrons. Modeling of the electron localization, and thus any redox process involving plutonium, is a complex task. Conventional density functional schemes that apply the local density approximation (LDA) or the generalized gradient approximation (GGA) underestimate the strong onsite Coulomb repulsion of the Pu 5f electrons and consequently fail to capture the correlation-driven localization. Therefore, the 5f electrons in elemental Pu, as well as in Pu compounds, require special attention. One promising way to improve contemporary LDA and GGA approaches is to modify the intra-atomic Coulomb interaction through the so-called LDA+U or GGA+U approach. in which the underestimation of the intraband Coulomb interaction is corrected by the Hubbard U parameter<sup>3,4,5</sup>. This method has been used to discuss the equilibrium lattice parameter of bulk Pu in Ref. 6,7,8. The choice of U is, however, not unambiguous and it is not trivial to determine its value a priori, though there are attempts to extract it from standard first-principles calculations. Hence, U is often fitted to reproduce a certain set of experimental data such as band gaps and structural prop-

In this paper we use the LDA+U and GGA+U schemes due to Dudarev et al.<sup>9</sup> to calculate the lattice parameters, electronic structure, and thermodynamic properties

of PuO<sub>2</sub> and Pu<sub>2</sub>O<sub>3</sub>. We discuss how these properties are affected by the choice of U as well as the choice of exchange-correlation potential, i.e., the LDA or the GGA, and how redox processes occurred in plutonium oxide can be explored in the LDA+U and GGA+U formalism. In addition, we notice that recently there have occurred a few experimental  $^{10,11,12}$  and theoretical  $^{11,13,14,15}$ studies of the electronic structures of plutonium oxides. In this paper we have compared our calculated  $\mathrm{LDA}/\mathrm{GGA} + U$  results with those reports. Our results show that while the pure LDA/GGA (without U correction) calculations fail to describe the ground-state behaviors of the plutonium oxides, such as the insulating nature, the magnetic configuration, and the 5f band gap, the present LDA/GGA+U approaches with tunable Coulomb parameters can effectively remedy those failures and the consequent results fit well in the attainable experimental data<sup>10,11,12</sup>.

This paper is organized as follows. The details of our calculations are described in Sec. II and in Sec. III we present and discuss the results. In Sec. IV, we summarize our findings.

### II. METHODOLOGY OF THE CALCULATION

The calculations were performed using the projector-augmented wave (PAW) method of Blöchl<sup>16</sup>, as implemented in the ab initio total-energy and molecular-dynamics program VASP (Vienna ab initio simulation program)<sup>17,18,19,20</sup>. PAW is an all-electron method that combines the accuracy of augmented-plane-wave methods with the efficiency of the pseudopotential approach. The PAW method is implemented in VASP with the frozen-core approximation. For the plane-wave set, a cut-off energy of 400 eV was used. The plutonium 6s, 6p, 7s, and 5f, and the oxygen 2s and 2p electrons were treated as valence electrons. The strong on-site Coulomb repulsion amongst the localized Pu 5f electrons are accounted for by using the formalism formulated by Dudarev et al.<sup>9</sup>. In this scheme the total LDA (GGA) energy functional

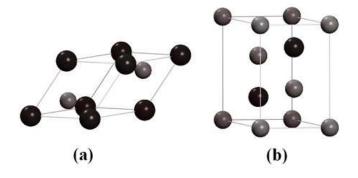


FIG. 1: (a) Unit cell of  $PuO_2$  containing 3 atoms. The black spheres are Pu atoms, the gray ones are oxygens. (b) Unit cell of  $Pu_2O_3$  containing five atoms.

is of the form

$$E_{\text{LDA(GGA)}+U} = E_{\text{LDA(GGA)}} + \frac{U - J}{2} \sum_{\sigma} \left[ \text{Tr} \rho^{\sigma} - \text{Tr} \left( \rho^{\sigma} \rho^{\sigma} \right) \right]$$
(1)

where  $\rho^{\sigma}$  is the density matrix of f states, and U and J are the spherically averaged screened Coulomb energy and the exchange energy, respectively. In this paper the Coulomb U is treated as a variable, while the exchange energy is set to be a constant  $J{=}0.75$  eV. This value of J is in the ball park of the commonly accepted one for  $\operatorname{Pu}^{6,21,22,23}$ . Since only the difference between U and J is significant<sup>9</sup>, thus we will henceforth label them as one single parameter, for simplicity labeled as U, while keeping in mind that the non-zero J has been used during calculations.

The exchange and correlation effects were treated in both the local density approximation and the generalized gradient approximation<sup>24</sup>. We studied PuO<sub>2</sub> in its ground-state fluorite structure (Fm3m) and the sesquioxide  $Pu_2O_3$  in the hexagonal  $\beta$ -type structure  $(P\bar{3}m1)$ . For PuO<sub>2</sub> we used a  $11 \times 11 \times 11$  Monkhorst-Pack k-point mesh<sup>25</sup> (56 irreducible k points) and for  $Pu_2O_3$  we used a  $9 \times 9 \times 6$  grid (57 irreducible k points). The electronic density of states (DOS) was obtained with  $15 \times 15 \times 15$  (120 irreducible k points) and  $11 \times 11 \times 9$  grid (120 irreducible k points) k-point meshes, respectively. The Brillouin-zone integration was performed using the modified tetrahedron method of Blöchl<sup>26</sup>. In order to study the reaction energy it is necessary to calculate the energy of an oxygen molecule  $(E_{O_2})$ . The effect of spin polarization was included in calculating  $E_{O_2}$ .

## III. RESULTS AND DISCUSSION

### A. Atomic and electronic structure of PuO<sub>2</sub>

Plutonium dioxide crystallizes in a CaF<sub>2</sub>-like ionic structure [Fig. 1(a)] with the plutonium and oxygen atoms forming face-centered and simple cubic sublattices,

respectively. In this arrangement each plutonium atom is located at the center of an oxygen cube, and for every four such cubes there is an empty one. In the ionic limit, formal charge for plutonium in PuO<sub>2</sub> is +4, corresponding to formal population of  $f^4$ . This leads to local S=2 plutonium moment, which can couple with other sites in either a ferromagnetic (FM) or antiferromagnetic (AFM) manner. PuO<sub>2</sub> is known to be an insulator<sup>27</sup> and some scattered experimental data<sup>28</sup> support the ground state of PuO<sub>2</sub> to be an AFM phase. In the present LDA/GGA+U approaches, we have considered the FM, AFM, and nonmagnetic phases for each choice of the value of U and then determined the groundstate phase by a subsequent total-energy comparison of these three phases. For  $PuO_2$ , at U=0, the ground state is a FM metal, which is in contrast to experiment. By increasing the amplitude of U, our LDA/GGA+U calculations correctly predicted an AFM insulating ground state. The turning value of U for this FM-AFM energy transition of the ground state is of  $\sim 1.5$  eV. In the discussion that follows, we therefore confined our report to the AFM solution for the  $PuO_2$ . A thorough description of the magnetic structure of plutonium oxides is beyond our intention in this paper, and we would like to leave it for the future studies.

The experimentally determined lattice parameter of  $PuO_2$  is  $a_0=5.396$  Åat  $25^{\circ}C^{29}$ . Here the calculated  $a_0$  and bulk modulus  $B_0$  of  $PuO_2$  were obtained from the corresponding energy minimization at constant volumes and by fitting a Murnaghan equation of state<sup>30</sup> to the resulting energy-volume data, respectively. The results as a function of U within the LDA and the GGA schemes are shown in Fig. 2(a) and (b) for  $a_0$  and  $B_0$ , respectively. For comparison, the experimental values of  $a_0^{29}$  and  $B_0^{31}$ are also shown in Fig. 2. For the pure DFT calculation (U=0), it shows in Fig. 2(a) that the LDA overbinds the compound and underestimates with respect to the experiment the lattice parameter by  $\sim 2\%$ , while the GGA calculation give a slight overestimate of  $a_0$ . After turning on the Hubbard U, one can see from Fig. 2(a) that for the LDA+U approach, although the lattice parameter is still underestimated in a wide range of U, the calculated  $a_0$  for PuO<sub>2</sub> improves upon the pure LDA by steadily increasing its amplitude with U. In fact, at a typical value of U=4 eV, the LDA+U gives  $a_0=5.36$  Å, which is very close to the experiment. On the other hand, with increasing U, the underbind effect brought about by the GGA+U is somewhat enlarged. As a comparison, at U=4 eV, the GGA+U gives  $a_0=5.47$  Å, which overestimates the experimental data by  $\sim 1.3\%$ . Overall, both the LDA+U and GGA+U results of the lattice parameter for the PuO<sub>2</sub> AFM phase are comparable with experiment by tuning in the calculations the Hubbard Uaround 4 eV. We have also calculated the equilibrium lattice parameter for the FM and nonmagnetic phases for  $PuO_2$ . The tendency of  $a_0$  with U for these two phases is similar to that for the present AFM phase. For the calculated bulk modulus  $B_0$  of the PuO<sub>2</sub> AFM phase, one

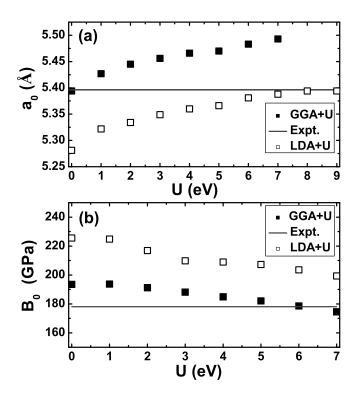


FIG. 2: Dependence of the lattice parameter (a) and bulk modulus (b) of  $PuO_2$  on U.

can see from Fig. 2(b) that its value varies with U over a rather broad range of 175 to 195 GPa for the GGA+Uand 200 to 230 GPa for the LDA+U. The LDA result of  $B_0$  is always higher than the GGA result, which is due to the above-mentioned "overbind" characteristics of the LDA approach. For the measurements of the equilibrium bulk modulus, there are no consistent results to date for the AFM PuO<sub>2</sub>. Here we compared our calculation to the experimental result of  $B_0=178$  GPa reported in Ref.<sup>31</sup>. One can see from Fig. 2(b) that the discrepancy between the present calculation and the experiment is most distinct at U=0. Both the LDA and the GGA give an overestimate, with the latter more close to the experimental data. By turning on the effective Coulomb interaction, the amplitude of  $B_0$  begins to decrease. At a typical value of U=4 eV, the LDA+U gives  $B_0=208$  GPa while the GGA+U gives  $B_0=184$  GPa. We notice that the recent hybrid density-funtional calculations<sup>14</sup> predict the bulk modulus to be 220 GPa for the antiferromagnetic PuO<sub>2</sub>, comparable with the present pure LDA results. For lattice parameter  $a_0$  it was predicted to be 5.46 in Ref. 14. To conclude (Fig. 2), comparing with the experimental data and the recent hybrid density-functional results, the accuracy of our atomic-structure prediction for the antiferromagnetic PuO<sub>2</sub> is quite satisfactory by tuning the effective Hubbard parameter U in a range 3-4 eV within the LDA/GGA+U approaches.

Besides the prominent changes in the atomic-structure parameters, the most dramatic improvement brought by

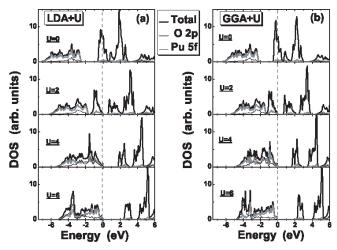


FIG. 3: The total DOS for the  $PuO_2$  antiferromagnetic phase computed in the (a) LDA+U and (b) GGA+U formalism with four selective values of U. The projected DOS for the  $Pu\ 5f$  and  $O\ 2p$  orbitals are also shown. The Fermi level was set to be zero.

the LDA/GGA+U when compared to the pure ones is in the description of electronic-structure properties. For this we have investigated the band structures of the PuO<sub>2</sub> AFM phase with the aim at seeing the fundamental influence by the inclusion of the on-site Coulomb interaction. The resultant total density of states (DOS) for four selective values of U are plotted in left (LDA+U) and right (GGA+U) panels in Fig. 3. For more clear illustration, the projected DOS for the Pu 5f and O 2p orbitals are also shown in Fig. 3. The Fermi energy  $E_F$  has been set to be zero. Without accounting for the on-site Coulomb repulsion (U=0), one can see that both two pure DFT methods predict an incorrect metallic ground state by non-zero occupation of Pu 5f states at  $E_F$ . When switching on U, as shown in Fig. 3, the Pu 5f band begins to split at  $E_F$  and tends to open a gap  $\Delta$ . The amplitude of this insulating gap increases with increasing U, see Fig. 4. Overall the LDA+U and GGA+U give an equivalent description of the one-electron behaviors in a wide range of U. At a typical value of U=4 eV, one can see from Fig. 3 that the occupied DOS is featured by two well-resolved peaks. The narrow one near -2.0 eV is principally Pu 5fin character, while the broad one near -4.0 eV is mostly O 2p. These two pronounced peaks have been observed in the recent photoemission measurements<sup>10,12</sup>. In addition, by increasing the amplitude of U, one prominent feature occurred in Fig. 3 is the increasing hybridization between Pu 5f and O 2p occupied states. This interesting mixing effect disappears in the cases of  $Pu_2O_3$  (see Fig. 6 below) and  $UO_2^{32}$ , for which the Pu (U) 5fand O 2p occupied bands are well separated. The presence of Pu(5f)-Pu(2p) hybridization in  $PuO_2$  implies a more covalent and stronger metal-ligand mixing than in  $Pu_2O_3$  and  $UO_2$ . This phenomenon appears surprising,

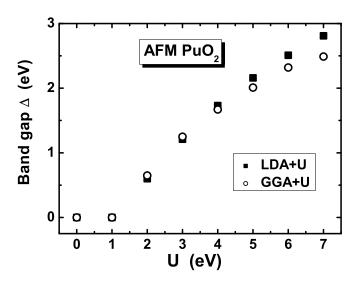


FIG. 4: The insulating band gap of the  $PuO_2$  antiferromagnetic phase as a function of U for the LDA (filled squares) and the GGA (hollow circles).

given the smaller overlap anticipated in Pu because of the smaller radius of the Pu 5f orbital. Experimentally, Butterfield et al. 10 and Gouder et al. 12 have reported the thin-film photoemission data for PuO<sub>2</sub>. The present overall picture which emerges from the LDA/GGA+Uwith properly selective Coulomb repulsion appear to be in satisfactory agreement with experiment. We have also compared our results given in Fig. 3 with the most recent calculations by Prodan et al. 14 based on newly developed screened Coulomb hybrid density functional. The agreement between our LDA/GGA+U (with  $U \sim 4 \text{ eV}$ ) results and those in Ref. <sup>14</sup> is also apparent. Interestingly, the above-mentioned orbital (Pu 5f and O 2p) mixing effect in PuO<sub>2</sub> has also been theoretically predicted by Prodan et al. 14,15, who hypotheses that the expected stabilization of the Pu 5f orbital energy relative to U 5f leads to an "accidental" degeneracy between the Pu 5f and O 2p levels, which in the first-order perturbation theory results in a higher degree of covalency regardless of small radius of the Pu 5 f orbital. Therefore, although the pure LDA and GGA fail to depict the electronic structure, especially the insulating nature and the occupiedstate character of PuO<sub>2</sub>, our present results show that by tuning the effective Hubbard parameter in a reasonable range, the LDA/GGA+U approaches will prominently improve upon the pure LDA/GGA calculations and thus can provide a satisfactory qualitative electronic structure description comparable with experiments and the hybrid DFT calculation. By further increasing U to 6 eV, one can see that the peak near -2.0 eV becomes weak and is mostly O 2p, while the peak near -4.0 eV becomes stronger and consists equally of Pu 5f and O 2p orbital. This picture of DOS is no longer valid since the peak near -2.0 eV has been confirmed to be due to the Pu 5f contribution. Thus the LDA/GGA+U approaches with U

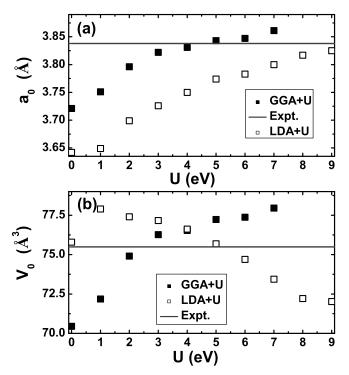


FIG. 5: Dependence of the equilibrium lattice parameter  $a_0$  (a) and the volume  $V_0$  of unit cell (b) of  $Pu_2O_3$  on U.

as large as 6 eV fails to describe the electronic structure of  $PuO_2$ .

### B. Atomic and electronic structure of Pu<sub>2</sub>O<sub>3</sub>

 $Pu_2O_3$  is an insulating oxide of the hexagonal  $\beta$ type  $(P\bar{3}m1)$  [Fig. 1(b)] with space group no. 164, the only phase of the sesquioxide that has been prepared with stoichiometric composition. Both magnetic susceptibility<sup>33</sup> and neutron diffraction<sup>34</sup> measurements have found Pu<sub>2</sub>O<sub>3</sub> to have an AFM structure at temperatures below 4.2 K, with the Pu moments  $\mu$  confined along the **z** axis in a simple +-+- alternation of spins. As with PuO<sub>2</sub>, we have considered the FM, AFM, and nonmagnetic phases and then determined the groundstate phase by comparing the equilibrium total energies of these three phases. At U=0, the calculated ground state is as for PuO<sub>2</sub> an incorrect FM metal. By increasing the amplitude of U, our LDA/GGA+U approaches correctly predicted the  $\beta$ -Pu<sub>2</sub>O<sub>3</sub> to be in an AFM insulating phase. The FM-AFM energy crossing occurs at a small U of  $\sim 1.5$  eV. We report in what follows on the  $Pu_2O_3$  AFM phase.

The calculated equilibrium lattice parameter  $a_0$  of  $Pu_2O_3$  is plotted in Fig. 5(a) as a function of U. It reveals that the relation between  $a_0$  and U does not follow a simple monotonic function. The turning point is at U=3 eV, below which  $a_0$  goes up rapidly with U. After

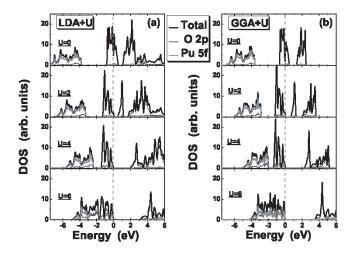


FIG. 6: The total DOS for the  $Pu_2O_3$  antiferromagnetic phase computed in the (a) LDA+U and (b) GGA+U formalism with four selective values of U. The projected DOS for the  $Pu\ 5f$  and  $O\ 2p$  orbitals are also shown.

crossing this turning point, the increase of  $a_0$  begins to slow down with U. Thus the curvature of  $a_0$  for small values of U is more significant than for large values of U. The decrease in curvature at large U corresponds to the separation of the occupied Pu 5f band from the unoccupied part, i.e., the transition from a metallic to an insulating ground state of Pu<sub>2</sub>O<sub>3</sub> (see below). This feature in the increase of  $a_0$  as a function of U is almost the same for the LDA and the GGA. The experimental data<sup>33,34</sup> of  $a_0$ =3.841 Åis well fitted at U=4 eV for the GGA, while the LDA always slightly underestimates  $a_0$ . Another feature shown in Fig. 5(a) is that at small values of U below 4 eV the GGA underestimates  $a_0$ , which is contrary to the general experience that in most cases (as shown for PuO<sub>2</sub>), the GGA often gives a slight overestimate of lattice parameter. This rarely-occurred feature may be due to the appearance of the other lattice parameter in  $\beta$ -Pu<sub>2</sub>O<sub>3</sub>, i.e., the rario  $c_0/a_0$  for the hexagonal crystalline structure. The equilibrium volume  $V_0$  of of the Pu<sub>2</sub>O<sub>3</sub> unit cell (including 5 atoms) as a function of U is plotted in Fig. 5(b). The experimental result<sup>33,34</sup> of  $V_0$  is also given for comparison. Although the tendency of  $V_0$  with U is remarkably opposite for the two DFT+U methods, the results mostly overlap at a typical value of U=4 eV, at which insulating gap for the  $Pu_2O_3$ is well formed. The different tendency of  $V_0$  with respect to U for the LDA and GGA may come from sensitivity of the anisotropy in Pu 5f orbitals to the treatment of the exchange-correlation potential. Combining Fig. 5(a) and (b) it is expected that both the LDA and the GGA may give a satisfactory prediction of the ground-state atomic structure for the  $Pu_2O_3$  by tuning U to be near 4 eV.

The LDA/GGA+U total DOS for the Pu<sub>2</sub>O<sub>3</sub> AFM phase are shown in Fig. 6 for four selective values of U. The projected DOS for the Pu 5f and O 2p orbitals

are also plotted. Both the LDA and GGA predict an incorrect metallic ground state for  $Pu_2O_3$  at U=0 by the presence of non-zero occupation of Pu 5f state at the Fermi energy  $E_F$ . When turning on the on-site Coulomb repulsion, the Pu 5f band begins to split and form an insulating gap  $\Delta$  at a critical value U=1 eV. The gap  $\Delta$  becomes large with increasing U, as shown in Fig. 7, from which one can see that the amplitude of  $\Delta$  for  $Pu_2O_3$  is almost equivalent to that for  $PuO_2$  at low U. At a typical value of U=4 eV, it reveals in Fig. 6 that the occupied DOS is featured by two peaks. The narrow one near -1.5 eV is principally Pu 5f in character, while the broad one around -4.0 eV is mostly O 2p. It is encouraging that these two pronounced peaks, as well as the overall appearance of the total DOS spectrum, fit well in recent photoemission experiments 10,12 on Pu<sub>2</sub>O<sub>3</sub>. We have also compared our results given in Fig. 6 with the recent calculations by Prodan et al. 14 using the hybrid density functional. Our LDA/GGA+Uresults (with  $U \sim 4 \text{ eV}$ ) for the Pu<sub>2</sub>O<sub>3</sub> AFM phase are in excellent agreement with those in Ref. 14. Unlike in  $PuO_2$ , the  $Pu ext{ } ext{5} ext{ } ext{and } ext{O} ext{ } ext{2} ext{p} ext{ states in } ext{Pu}_2O_3 ext{ are well sep$ arated in the DOS spectrum. This feature is similar to that of UO<sub>2</sub><sup>32</sup>, which also exhibits two distinct peaks of U 5f and O 2p parentage. Remarkably, the similar trend has also been theoretically reported on Pu<sub>2</sub>O<sub>3</sub> in Ref.<sup>14</sup> within the hybrid-density-functional framework. A consistent explanation with the Pu(5f)-O(2p) hybridization in PuO<sub>2</sub> may sustain by understanding the orbital separation in Pu<sub>2</sub>O<sub>3</sub> as a consequence of the more weakly bound Pu 5f site energy associated with the less highly charged Pu<sup>3+</sup> ion<sup>13</sup>. With further increasing the effective intratomic Coulomb interaction to U=6 eV, as shown in Fig. 6, the separation of the Pu 5f from O 2p projected DOS is blurred by the increasing spectrum weight of the former around -4 eV, which overlaps largely with the O 2p. This no longer accord with the experiments<sup>10,12</sup>. Therefore, as with PuO<sub>2</sub>, the LDA/GGA+U approaches with U as large as 6 eV fail to describe the electronic structure of  $Pu_2O_3$ .

### C. Oxidation reaction energy

Oxidation of Pu<sub>2</sub>O<sub>3</sub> via the reaction

$$Pu_2O_3 + \frac{1}{2}O_2 \to 2PuO_2$$
 (2)

leads to formation of stoichiometric  $PuO_2$ . The dependence of the transformation reaction energy on U is presented in Fig. 8. One can see that both the LDA and the GGA show the same dependence of reaction energy on the on-site Coulomb interaction. That is, at small values of U which correspond to the metallic ground state for both  $PuO_2$  and  $Pu_2O_3$ , the reaction energy is independent of U. Above the metallic-insulating transition, our calculated reaction energy decreases linearly with increasing U. The reason for this behavior is that a high

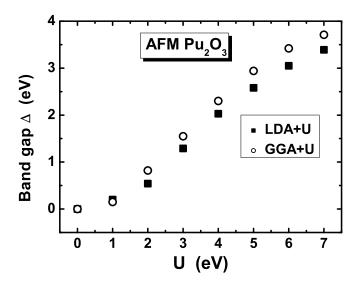


FIG. 7: The insulating band gap of the  $Pu_2O_3$  antiferromagnetic phase as a function of U for the LDA (filled squares) and the GGA (hollow circles).

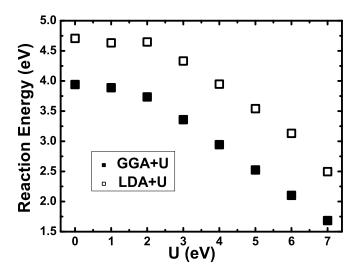


FIG. 8: Dependence of the  $Pu_2O_3 + \frac{1}{2}O_2 \rightarrow 2PuO_2$  reaction energy on U.

U favors localization and thus facilitates the transition. Density functional theory is known to overestimate the binding energy of  $O_2$  and this should result in an underestimation of the present reaction energy via the  $E_{O_2}$  term. Consequently we cannot expect a perfect agreement with experiments for the present reaction energy.

However, this error is independent of any conditions in the plutonium oxide and thus can be remedied by shifting the energy of  $O_2$  so as to give the experimental binding energy. In the LDA the  $O_2$  binding energy is overestimated by 1.2 eV/0.5  $O_2$  and in the GGA the corresponding number is 0.8 eV/0.5  $O_2$ . The GGA always predicts a lower value of the reaction energy, as seen from Fig. 9.

#### IV. CONCLUSIONS

We have studied the structural, electronic, and thermodynamic properties of the antiferromagnetic PuO<sub>2</sub> and  $Pu_2O_3$  within the LDA+U and GGA+U frameworks. The atomic structure, including lattice parameters and bulk modulus, and the one-electron behaviors of these kinds of plutonium oxides have been systematically investigated as a function of the effective on-site Coulomb repulsion parameter U. We find that both the LDA+Uand GGA+U considerably improves upon the traditional density functionals, providing a first-principles description of plutonium oxides in satisfactory qualitative agreement with experiment. Also our present results are well comparable to those obtained through newly developed hybrid DFT method. Specially, from the LDA/GGA+Ustudy of the lattice parameter of PuO<sub>2</sub> we find that the experimental data of  $a_0$  can be gradually approached by steadly increasing U to be in an acceptable range around 4 eV. The incorrect metallic ground state at purely LDA or GGA (U=0) for both PuO<sub>2</sub> and Pu<sub>2</sub>O<sub>3</sub> can be readily corrected by a systematic inclusion of nonzero U, which forces the Pu 5f band to split at the Fermi level and thus drives the metallic-insulating transition. The insulating band gaps for PuO<sub>2</sub> and Pu<sub>2</sub>O<sub>3</sub> have been shown as a function of U. The oxidation reaction  $Pu_2O_3+0.5O_2 \rightarrow 2PuO_2$  has also been studied by systematically calculating the reaction energy as a function of U. Our results show that the oxidation process of the Pu<sub>2</sub>O<sub>3</sub> is an exothermic reaction, which is mostly responsible for the experimentally observed<sup>36</sup> plutonium pyrophoricity at 150°C-200°C. Also we have shown that above the metallic-insulating transition, the reaction energy decreases with increasing U for the LDA and the GGA schemes. We expect these calculated results are useful for the future studies on the surface oxidation and corrosion of metallic plutonium.

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<sup>&</sup>lt;sup>1</sup> J.M. Haschke, Los Alamos Science **26**, 253 (2000).

<sup>&</sup>lt;sup>2</sup> J.M. Haschke, T.H. Allen, and L.A. Morales, Science 287,

<sup>285 (2000).</sup> 

<sup>&</sup>lt;sup>3</sup> V.I. Anisimov, J. Zaanen, and O.K. Anderson, Phys. Rev. B 44, 943 (1991).

<sup>&</sup>lt;sup>4</sup> V.I. Anisimov, I.V. Solovyev, M.A. Korotin, M.T. Czyżyk,

- and G.A. Sawatzky, Phys. Rev. B 48, 16929 (1993).
- <sup>5</sup> I.V. Solovyev, P.H. Dederichs, and V.I. Anisimov, Phys. Rev. B **50**, 16861 (1994).
- <sup>6</sup> S.Y. Savrasov and G. Kotliar, Phys. Rev. Lett. **84**, 3670 (2000).
- <sup>7</sup> A.B. Shick, V. Drchal, and L. Havela, Europhys. Lett. **69**, 588 (2005).
- <sup>8</sup> A. Shick, L. Havela, J. Kolorenč, V. Drchal, T. Gouder, and P.M. Oppeneer, Phys. Rev. B 73, 104415 (2006).
- <sup>9</sup> S.L. Dudarev, G.A. Botton, S.Y. Savrasov, C.J. Humphreys, and A.P. Sutton, Phys. Rev. B **57**, 1505 (1998).
- M. Butterfield, T. Durakiewicz, E. Guziewicz, J. Joyce, A. Arko, K. Graham, D. Moore, and L. Morales, Surf. Sci. 571, 74 (2004).
- <sup>11</sup> M.T. Butterfield, T. Durakiewicz, I.D. Prodan, G.E. Scuseria, E. Guziewicz, J.A. Sordo, K.N. Kudin, R.L. Martin, J.J. Joyce, A.J. Arko, K.S. Graham, D.P. Moore, and L.A. Morales, Surf. Sci. 600, 1637 (2006).
- <sup>12</sup> T. Gouder, A. Seibert, L. Havela, and J. Rebizant, Surf. Sci. 601, L77 (2007).
- <sup>13</sup> I.D. Prodan, G.E. Scuseria, J.A. Sordo, K.N. Kudin, and R.L. Martin, J. Chem. Phys. **123**, 014703 (2005).
- <sup>14</sup> I.D. Prodan, G.E. Scuseria, and R.L. Martin, Phys. Rev. B **73**, 045104 (2006).
- <sup>15</sup> I.D. Prodan, G.E. Scuseria, and R.L. Martin, Phys. Rev. B **76**, 033101 (2007).
- <sup>16</sup> P.E. Blöchl, Phys. Rev. B **50**, 17953 (1994).
- <sup>17</sup> G. Kresse and J. Hafner, Phys. Rev. B **48**, 13115 (1993).
- <sup>18</sup> G. Kresse and J. Furthmüller, Comput. Mater. Sci. 6, 15 (1996).
- <sup>19</sup> G. Kresse and J. Furthmüller, Phys. Rev. B **54**, 11169 (1996).
- <sup>20</sup> G. Kresse and D. Joubert, Phys. Rev. B **59**, 1758 (1999).
- <sup>21</sup> D. van der Marel and G.A. Sawatzky, Phys. Rev. B **37**, 10674 (1988); J.F. Herbst, R.E. Watson, and I. Lindgren, *ibid.* **14**, 3265 (1976).

- <sup>22</sup> A.B. Shick, A. I. Liechtenstein, and W.E. Pickett, Phys. Rev. B **60**, 10763 (1999).
- <sup>23</sup> A.B. Shick, V. Janiš, and P.M. Oppeneer, Phys. Rev. Lett. 94, 016401 (2005).
- <sup>24</sup> J.P. Perdew, J.A. Chevary, S.H. Vosko, K.A. Jackson, M.R. Pederson, D.J. Singh, and C. Fiolhais, Phys. Rev. B 46, 6671 (1992).
- <sup>25</sup> H.J. Monkhorst and J.D. Pack, Phys. Rev. B **13**, 5188 (1976).
- <sup>26</sup> P.E. Blöchl, O. Jepsen, and O.K. Andersen, Phys. Rev. B 49, 16223 (1994).
- <sup>27</sup> C. E. McNeilly, J. Nucl. Mater. **11**, 53 (1964).
- P. Santini, R. Lémanski, and P. Erdős, Adv. Phys. 48, 537 (1999); M. Colarieti-Tosti, O. Eriksson, L. Nordström, J. Wills, and M.S.S. Brooks, Phys. Rev. B 65, 195102 (2002); S. Kern, R. A. Robinson, H. Nakotte, G. H. Lander, B. Cort, P. Watson, and F. A. Vigil, *ibid.* 59, 104 (1999); G. Raphael and R. Lallement, Solid State Commun. 6, 383 (1968).
- <sup>29</sup> R. G. Haire, J. M. Haschke, MRSBull. 689 (September 2001).
- <sup>30</sup> F.D. Murnaghan, Proc. Natl. Acad. Sci. U.S.A. **30**, 244 (1944).
- <sup>31</sup> M. Idiri, T. LeBihan, S. Heathman, and J. Rebizant, Phys. Rev. B **70**, 014113 (2004).
- <sup>32</sup> K.N. Kudin, G.E. Scuseria, and R.L. Martin, Phys. Rev. Lett. **89**, 266402 (2002).
- <sup>33</sup> B. McCart, G.H. Lander, and A.T. Aldred, J. Chem. Phys. 74, 5263 (1981).
- <sup>34</sup> M. Wulff and G.H. Lander, J. Chem. Phys. **89**, 3295 (1988).
- <sup>35</sup> F.H. Ellinger, The Metal Plutonium (The University of Chicago Press, Chicago, IL, 1961).
- <sup>36</sup> J.C. Martz, J.M. Haschke, and J.L. Stakebake, J. Nuclear Materials 210, 130 (1994).